
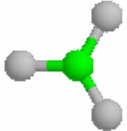




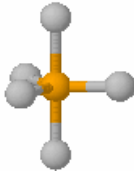
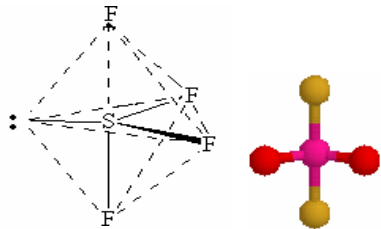
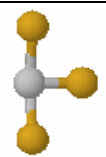

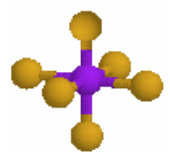
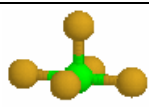


tipo		n° de pares electrónicos			Ejemplos			
Código	hibridación	totales	de enlace	sin compartir	Molécula o ión	No polares	polares	imagen
AX ₂	sp	2	2	0	lineal	HgCl ₂ BeH ₂ CO ₂	HCN	
AX ₃	sp ²	3	3	0	triangular plana	BF ₃	OCCl ₂	
AX ₂ E	sp ²	3	2	1	angular		O ₃ SO ₂ NO ₂ ⁻	
AX ₄	sp ³	4	4	0	tetraédrica	CH ₄	CH ₃ Cl	
AX ₃ E	sp ³	4	3	1	piramide trigonal		NH ₃	
AX ₂ E ₂	sp ³	4	2	2	angular		H ₂ O	
AX ₅	dsp ³	5	5	0	bipirámide trigonal	PCl ₅ PF ₅	PCl ₄ F	

AX_4E	dsp^3	5	4	1	tetraédrica irregular		SF_4 IF_4^+	
AX_3E_2	dsp^3	5	3	2	forma T		ClF_3	
AX_2E_3	dsp^3	5	2	3	lineal	XeF_2	$BrICl_2^-$	
AX_6	d^2sp^3	6	6	0	octaédrica		SF_6	
AX_5E	d^2sp^3	6	5	1	pirámide cuadrada		IF_5	
AX_4E_2	d^2sp^3	6	4	2	cuadrada plana	XeF_4		