

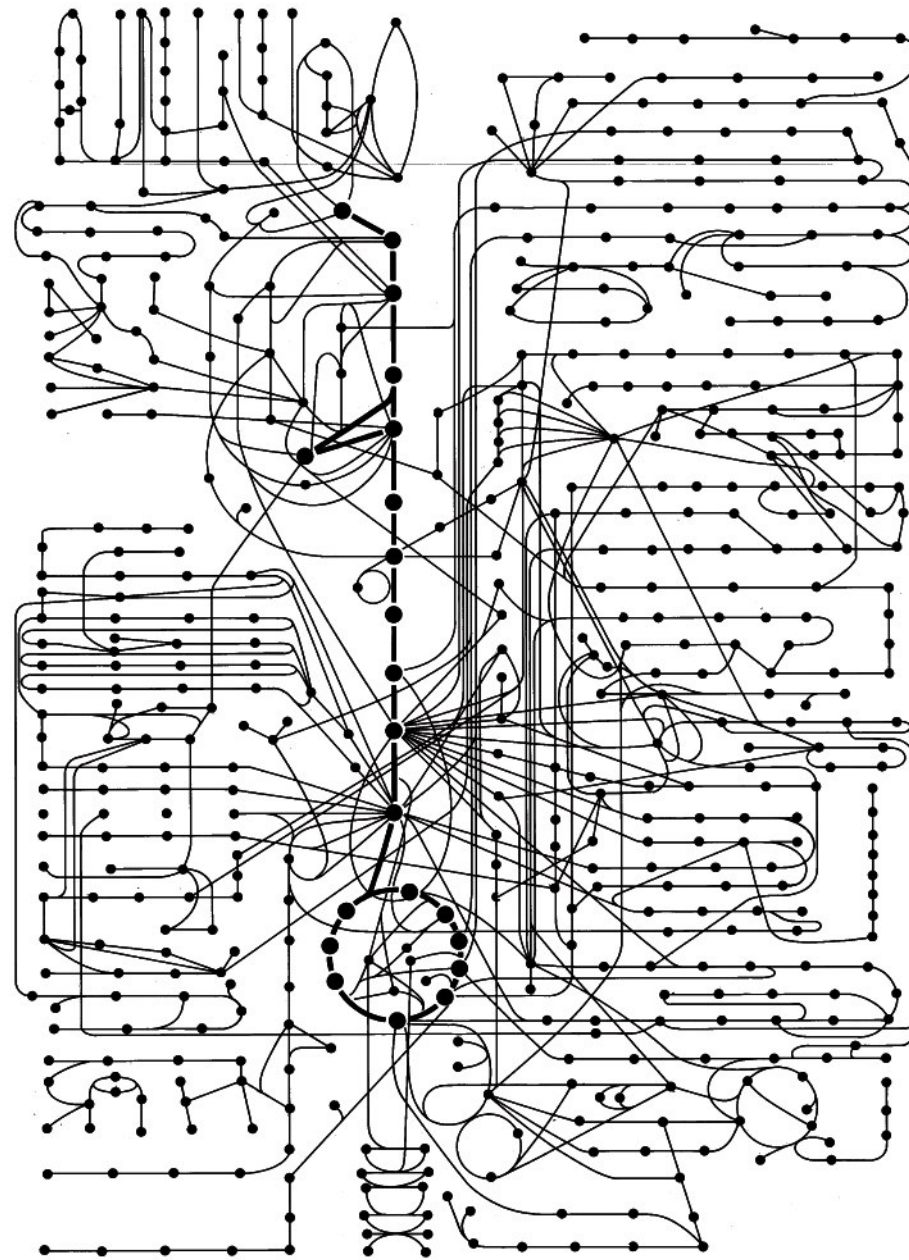
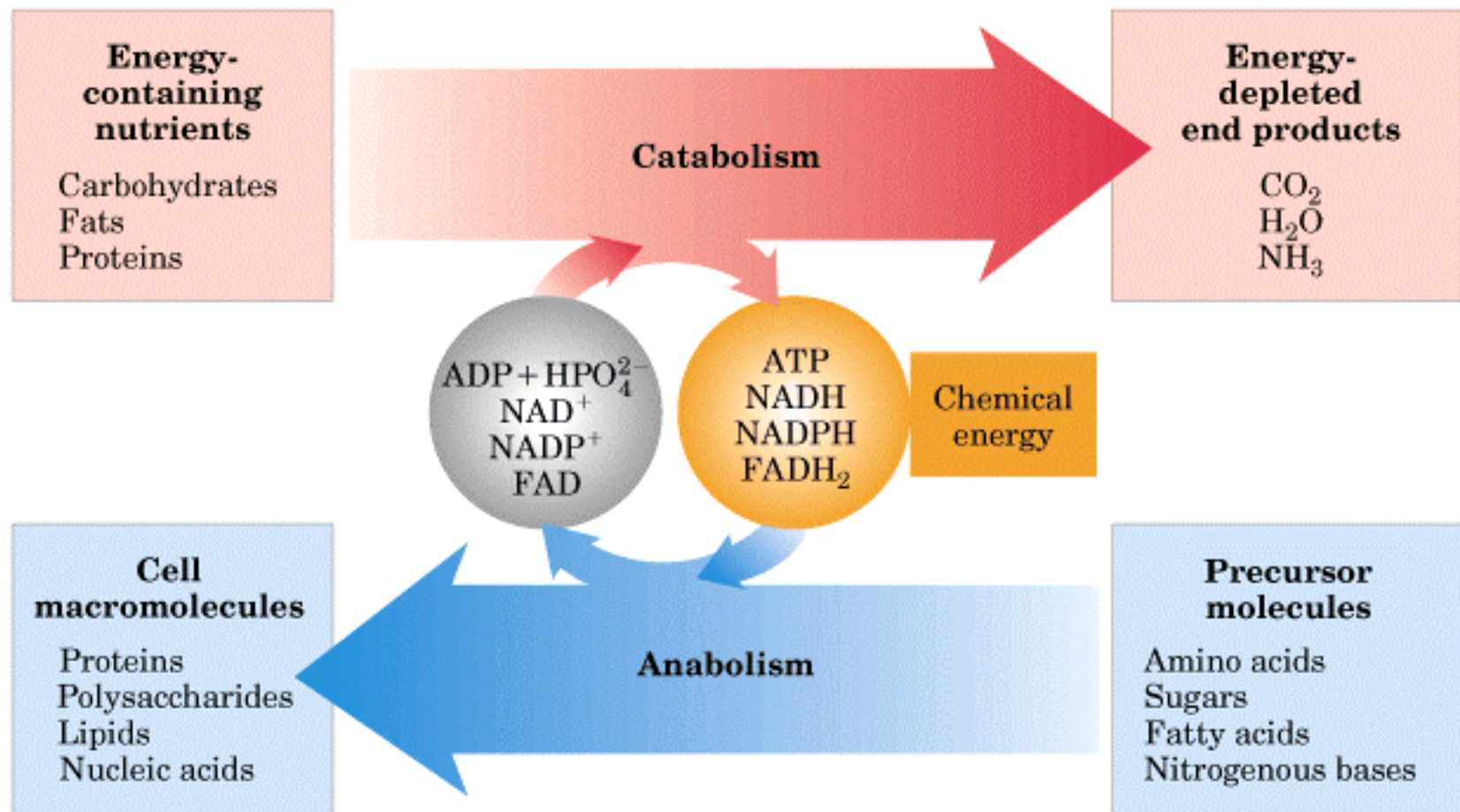
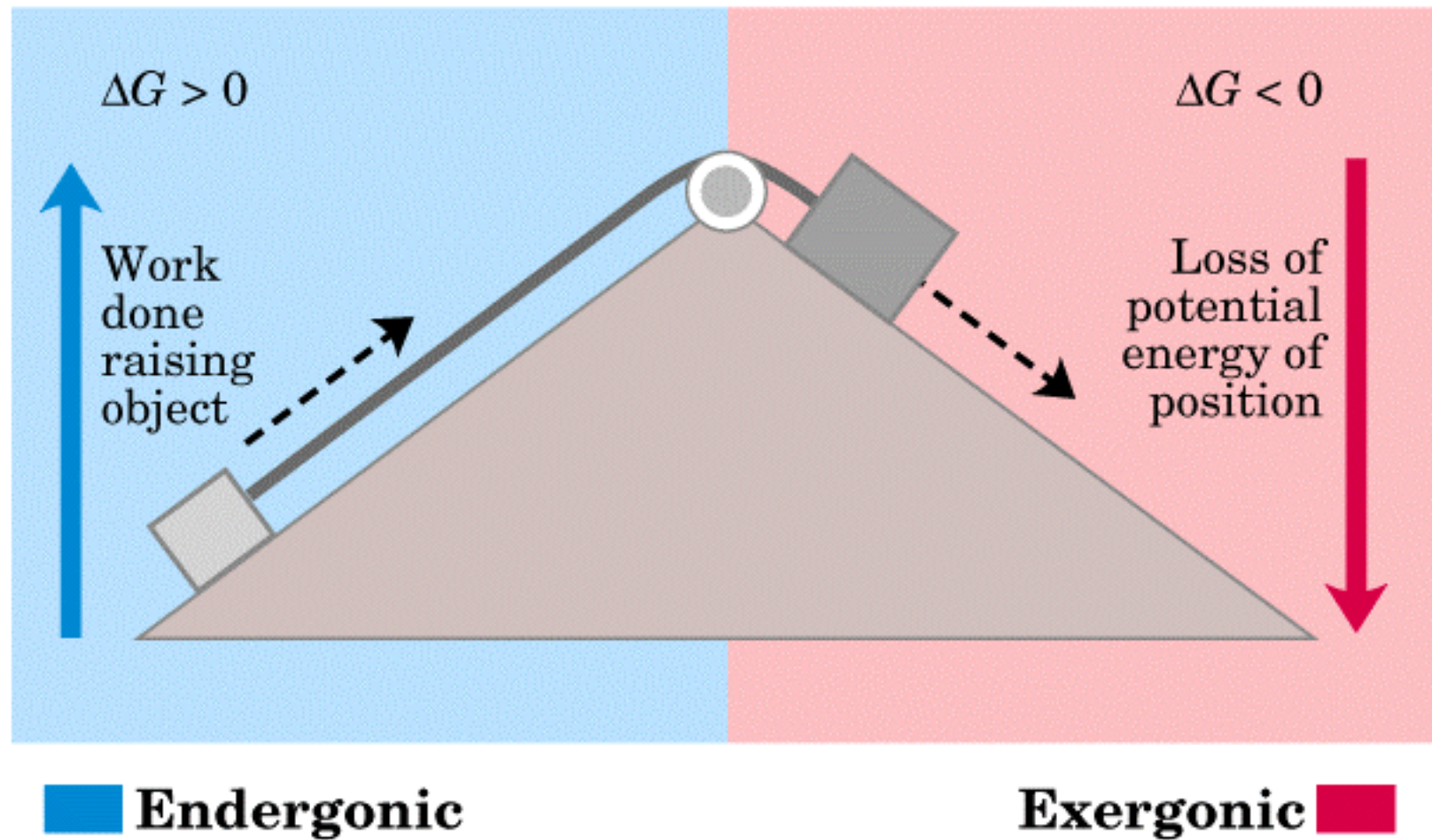


Figure 2-35 *Molecular Biology of the Cell* (© Garland Science 2008)



(a) Mechanical example

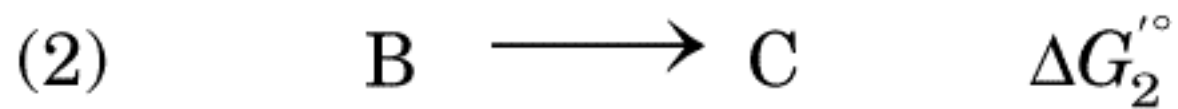


$$\Delta G'^{\circ} = -RT \ln K'_{\text{eq}}$$

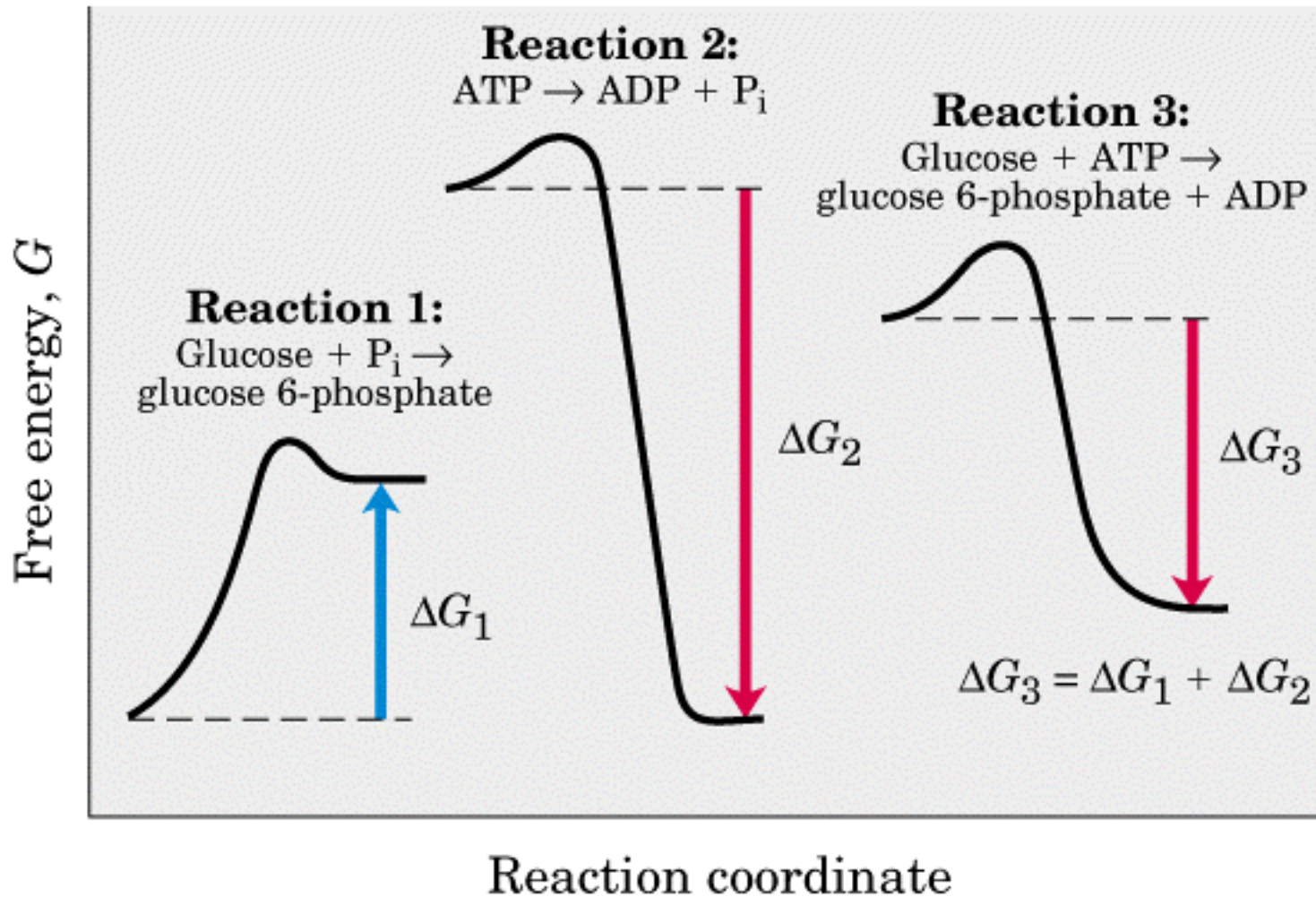
table 14–3

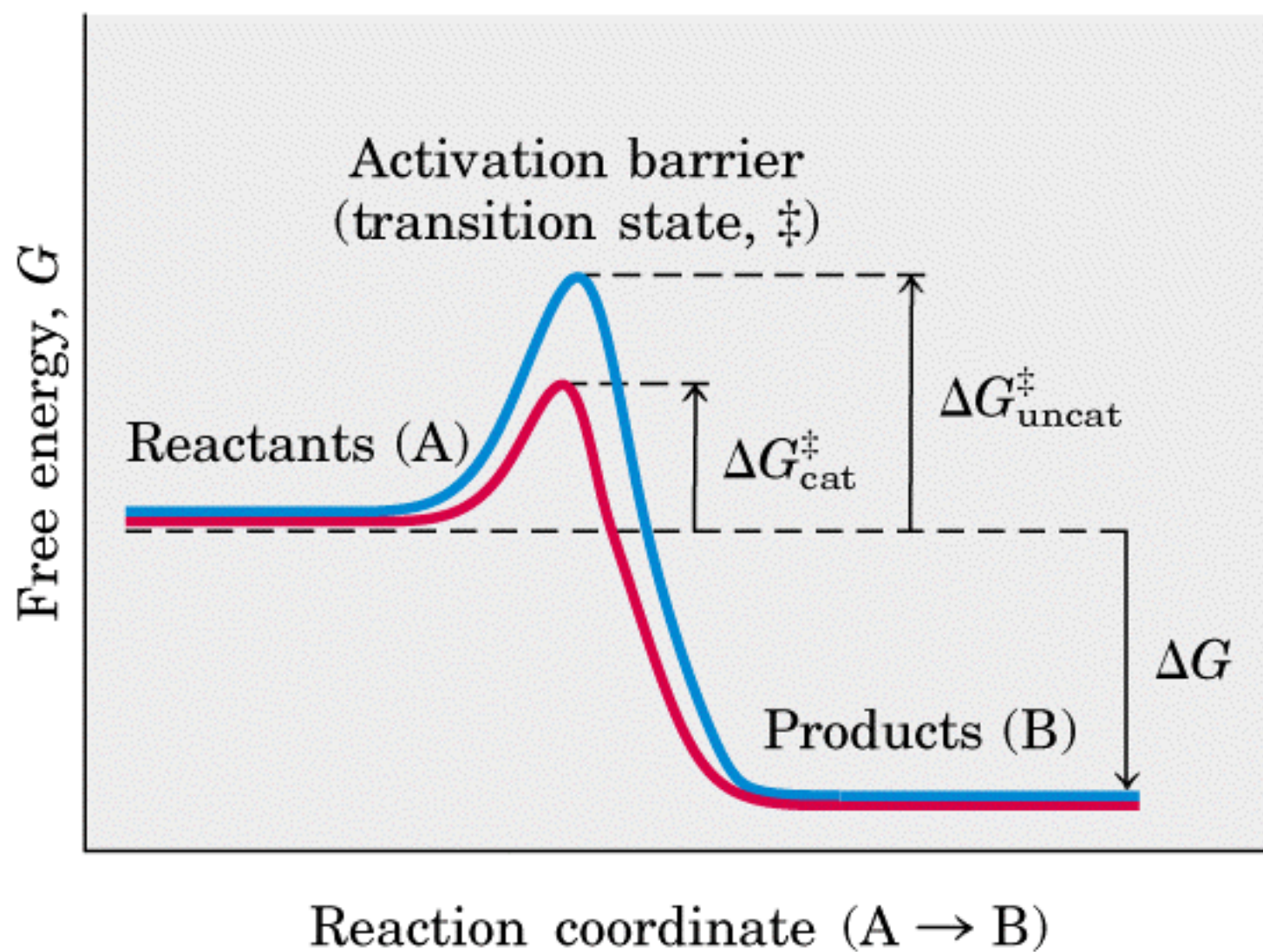
Relationships among K'_{eq} , $\Delta G'^{\circ}$, and the Direction of Chemical Reactions under Standard Conditions

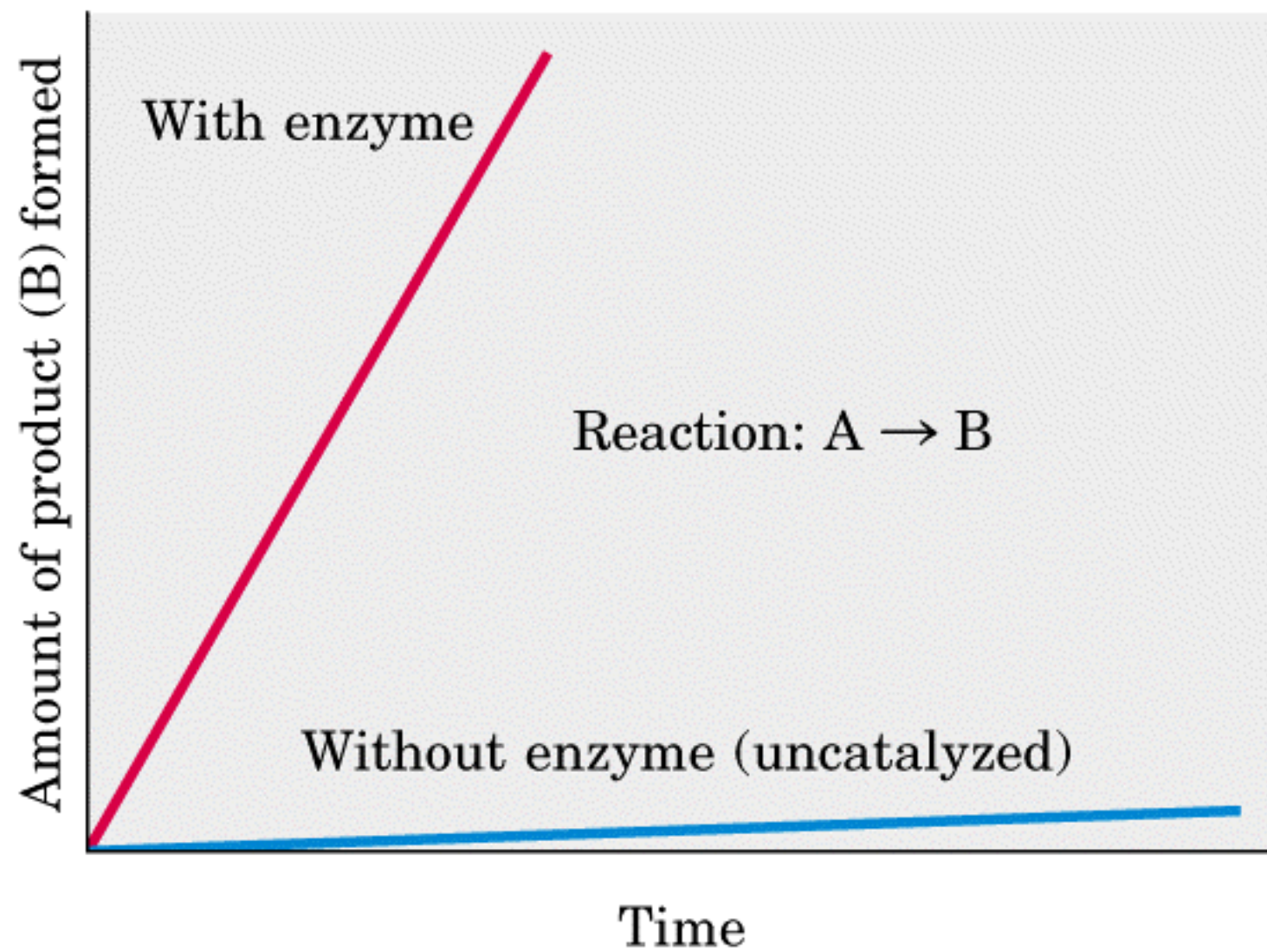
When K'_{eq} is	$\Delta G'^{\circ}$ is	Starting with 1 M components the reaction
>1.0	Negative	Proceeds forward
1.0	Zero	Is at equilibrium
<1.0	Positive	Proceeds in reverse

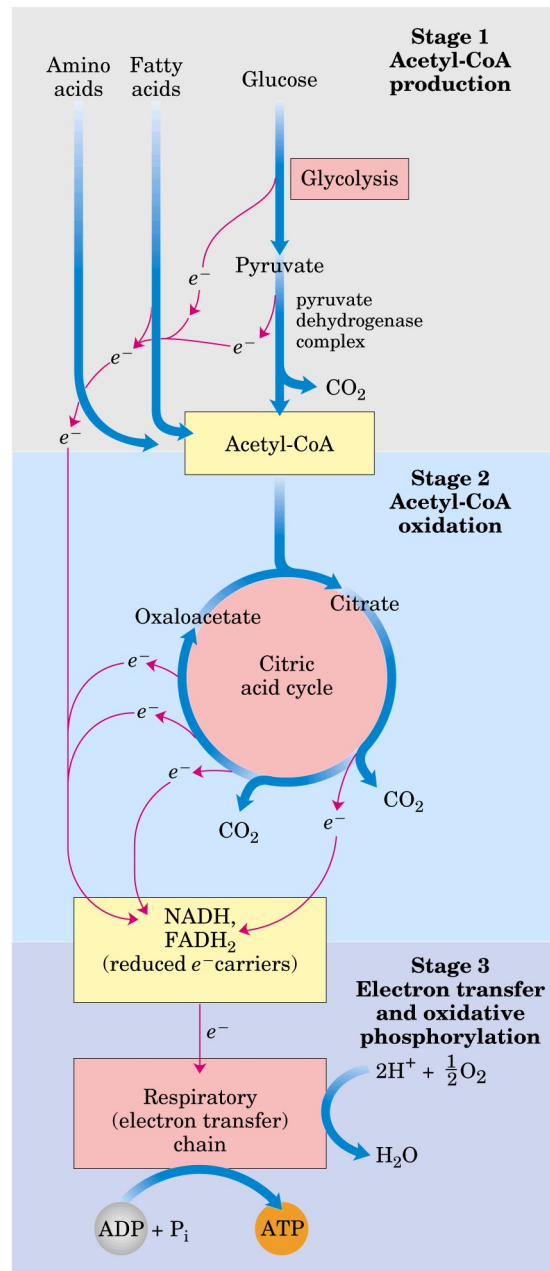


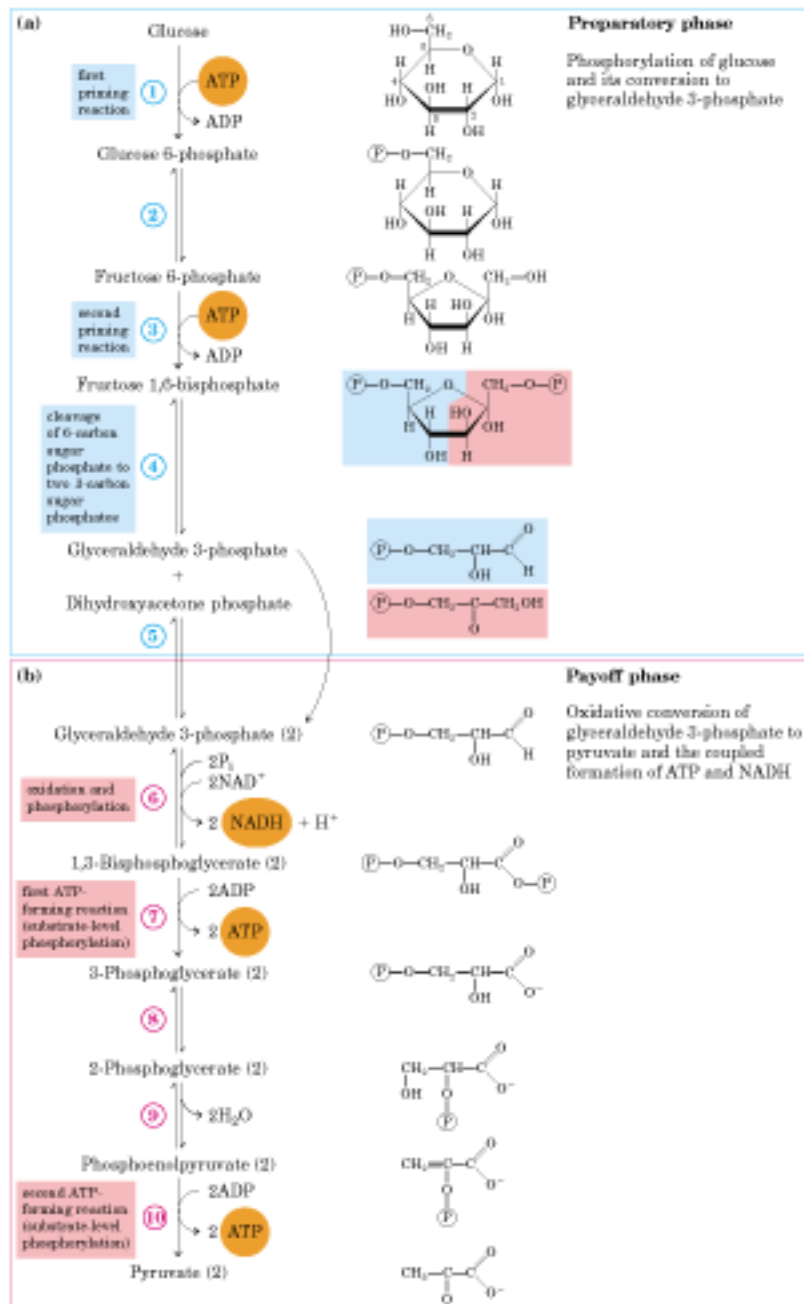
(b) Chemical example

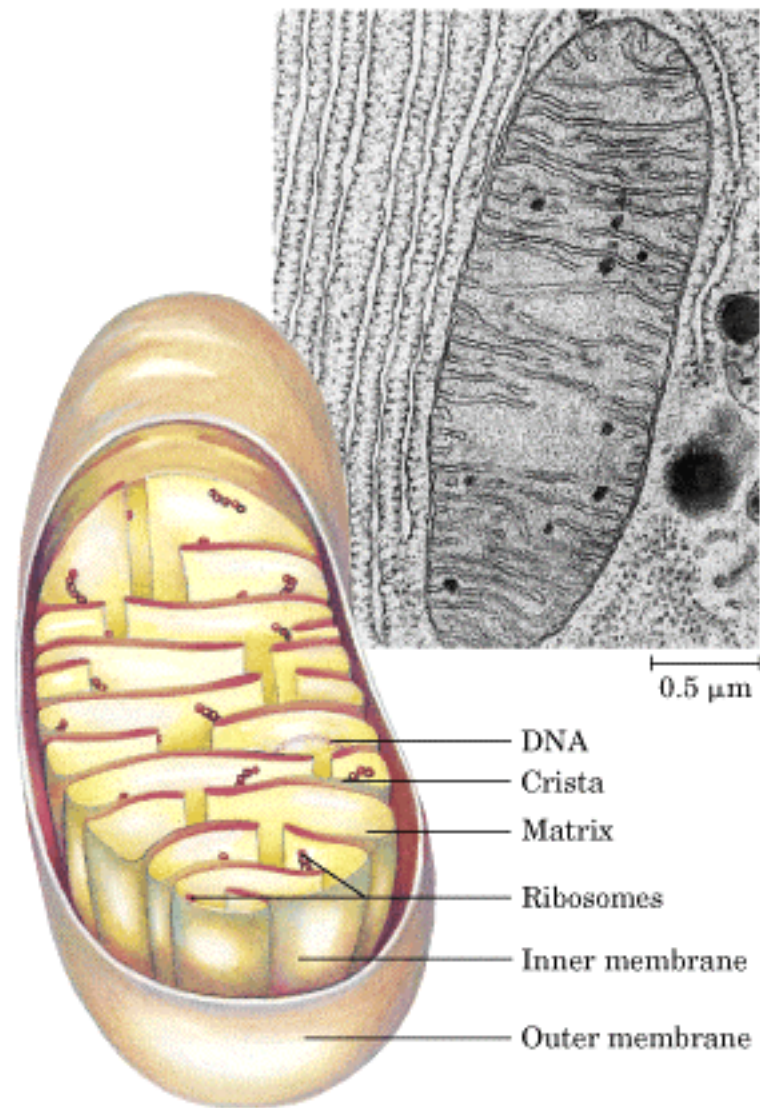


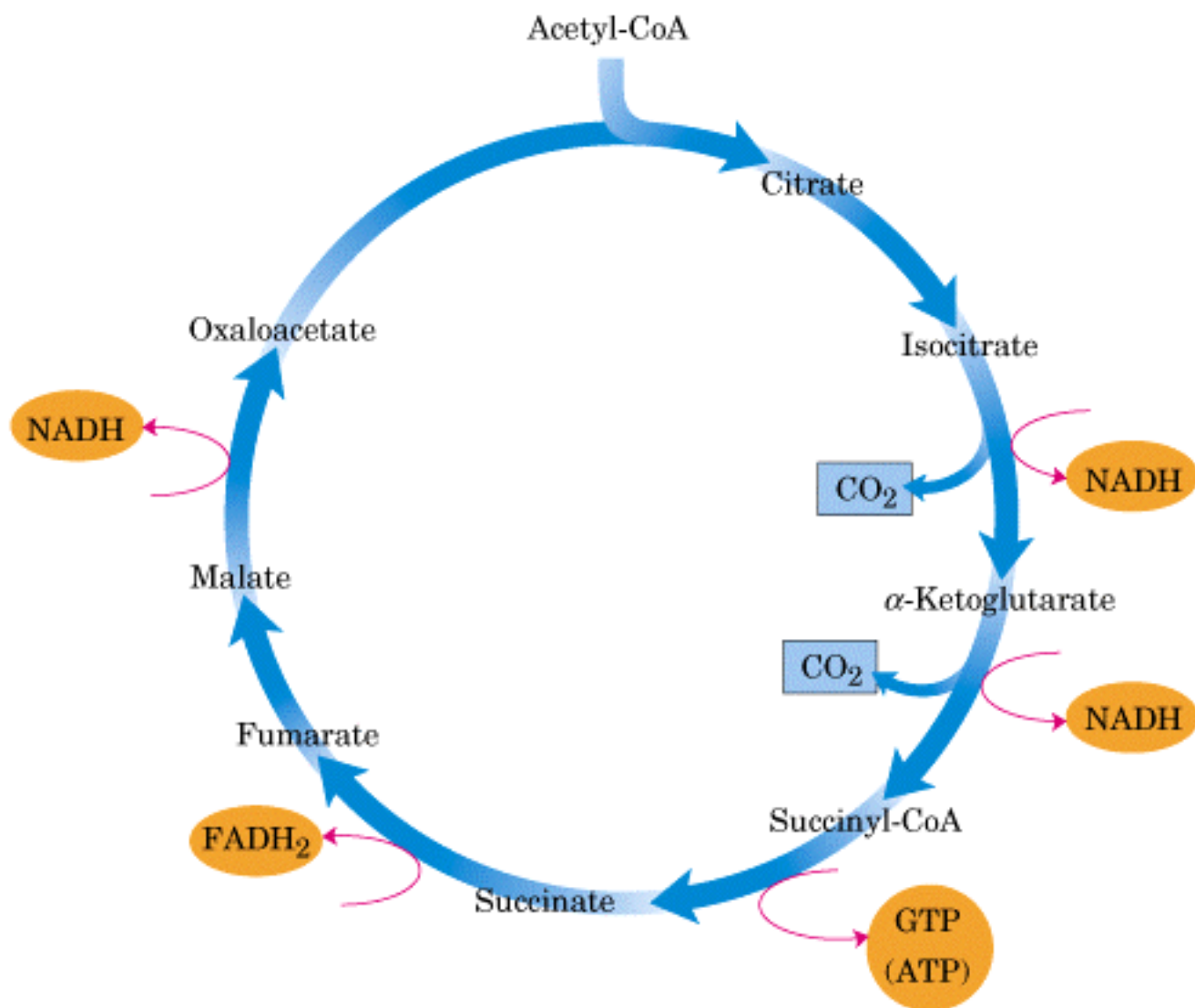












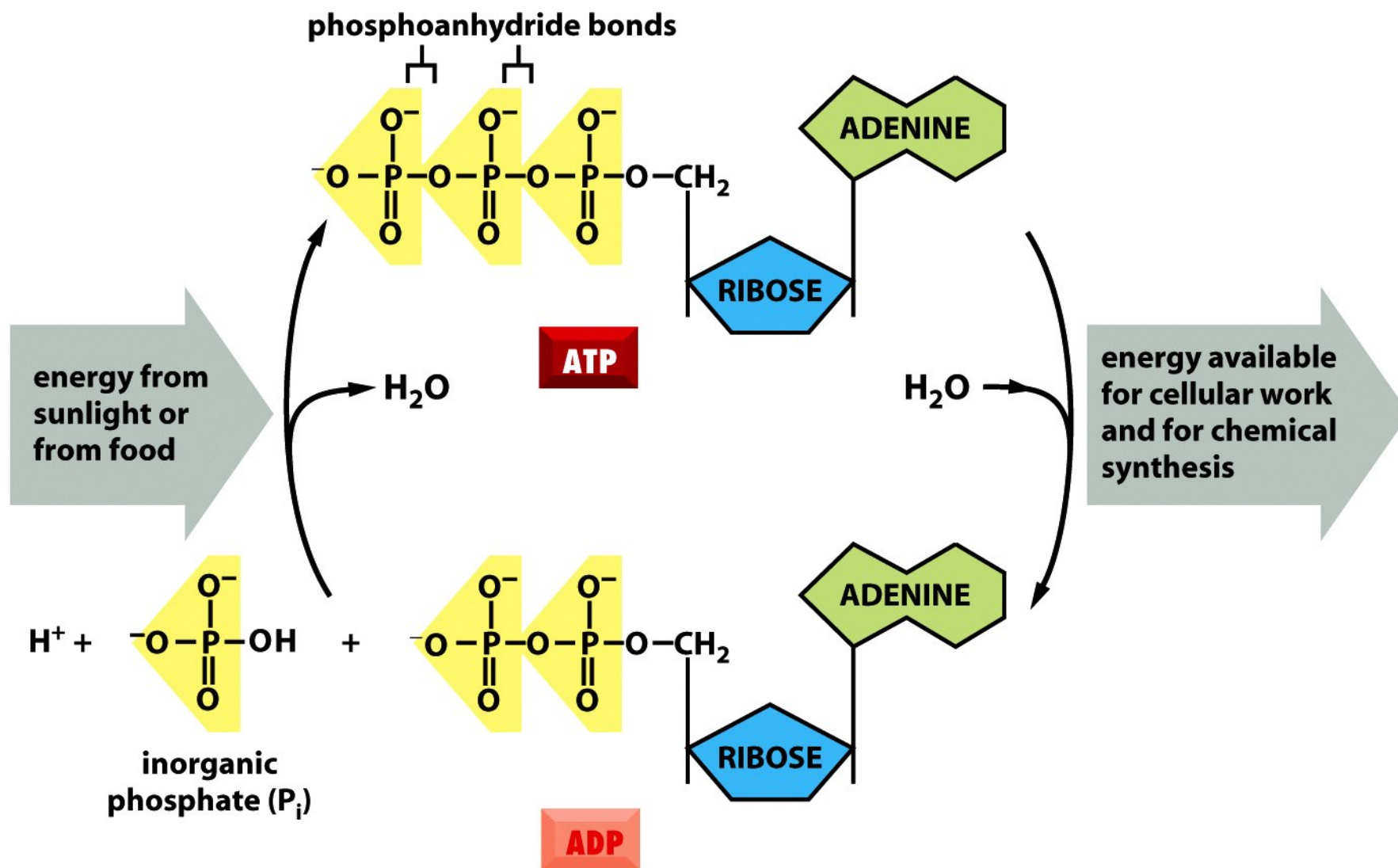
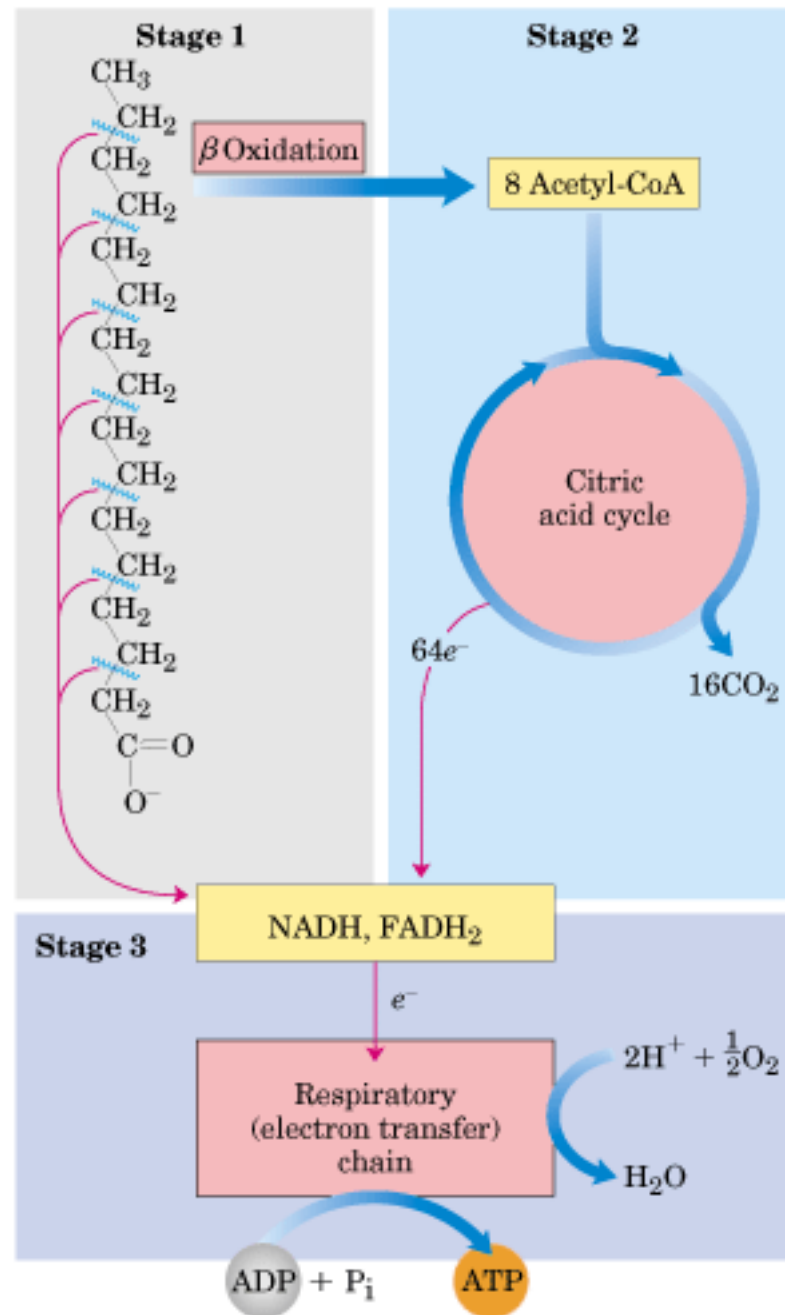
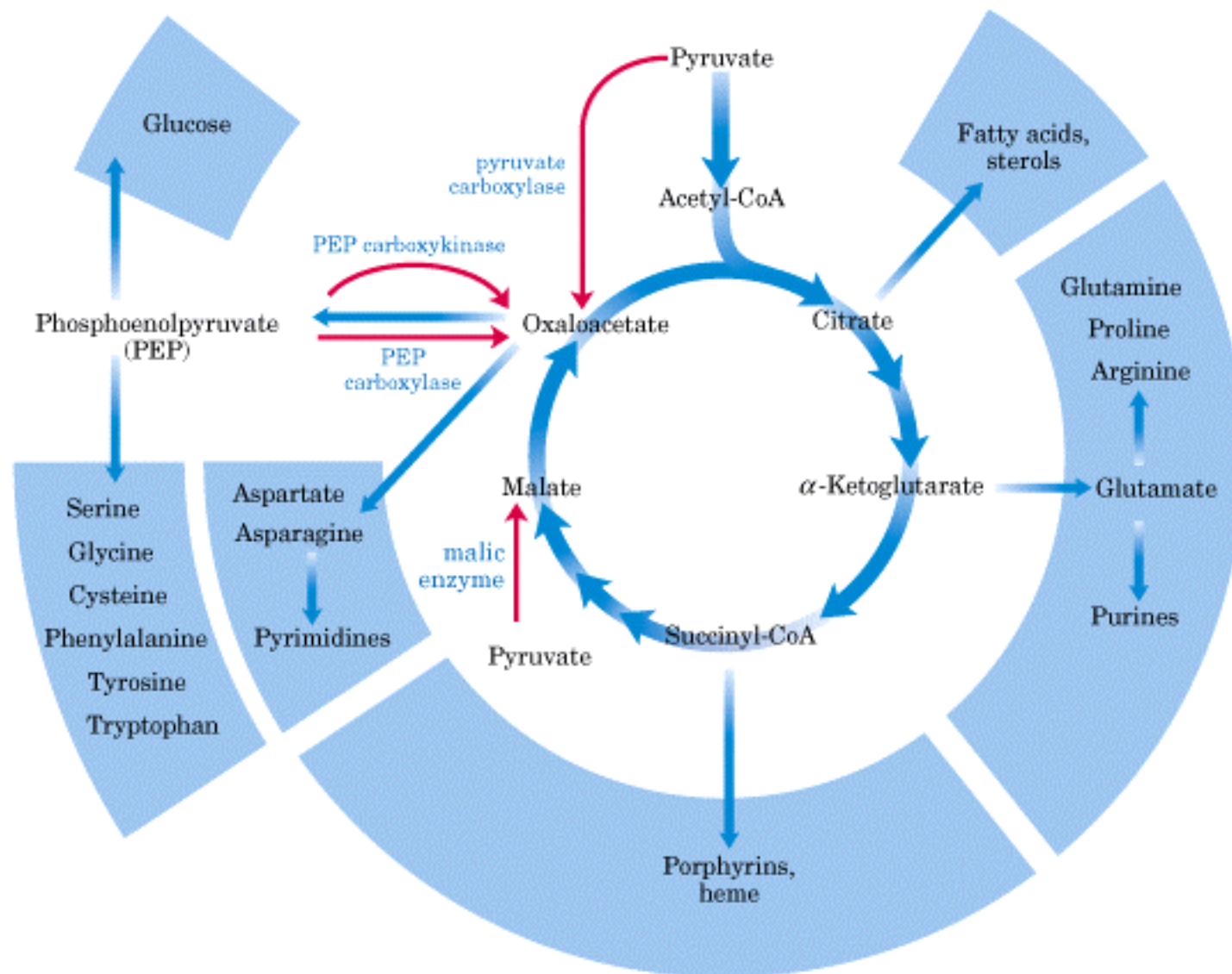
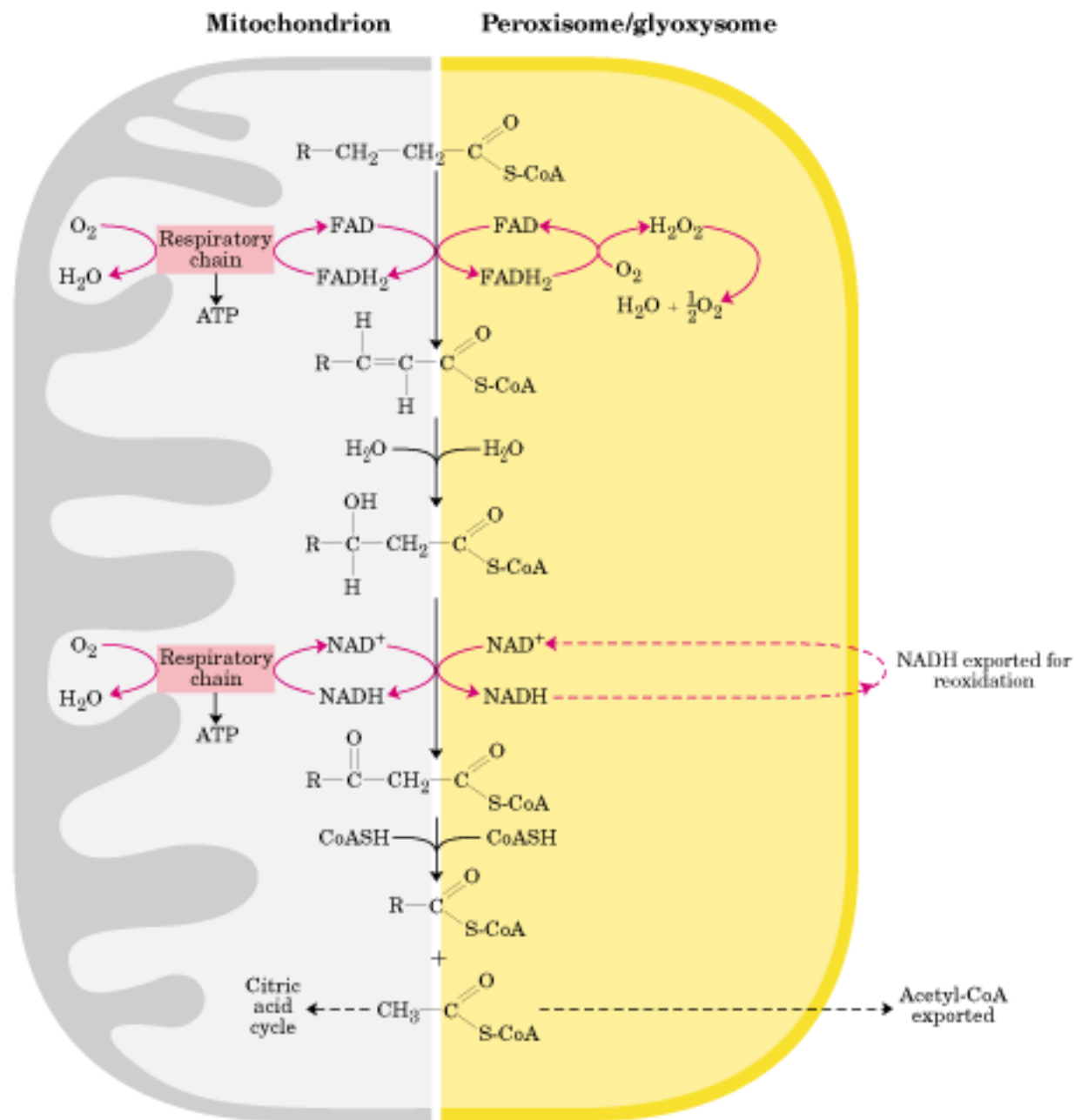


Figure 2-27 *Molecular Biology of the Cell* (© Garland Science 2008)







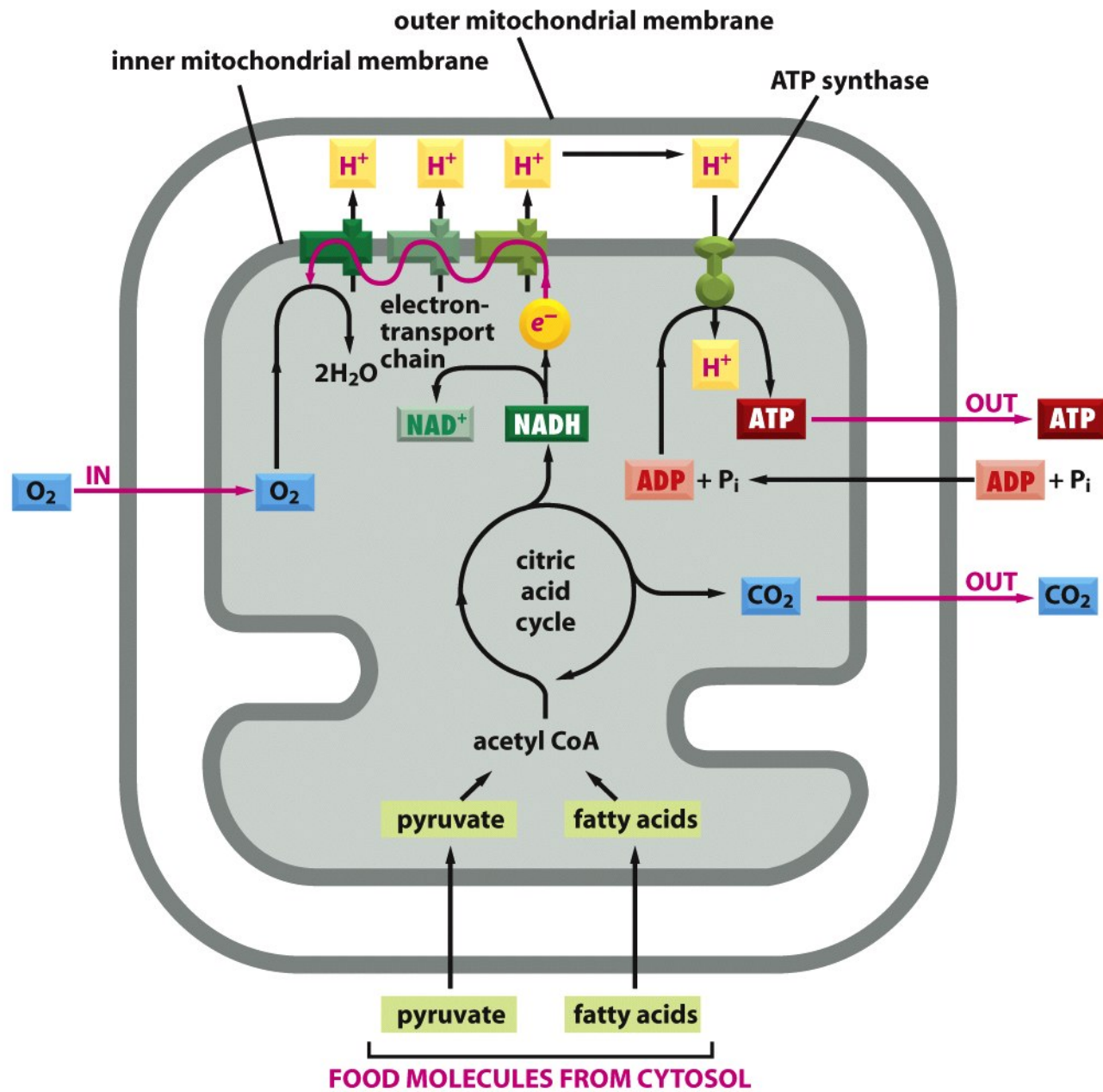
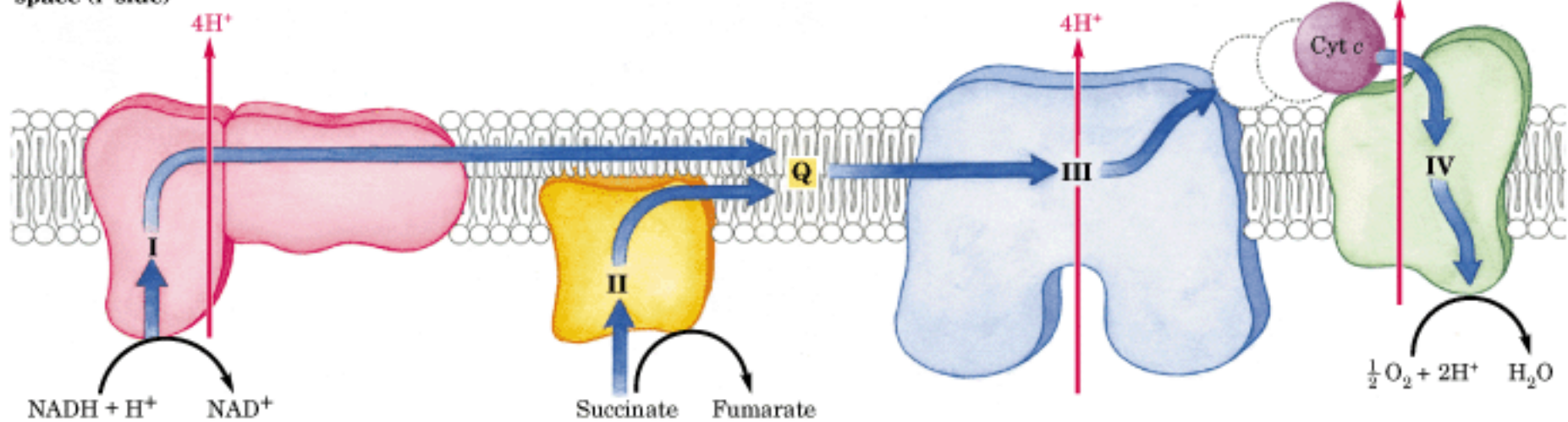


Figure 14-10 *Molecular Biology of the Cell* (© Garland Science 2008)

Intermembrane
space (P side)



Matrix (N side)

MITOCHONDRION

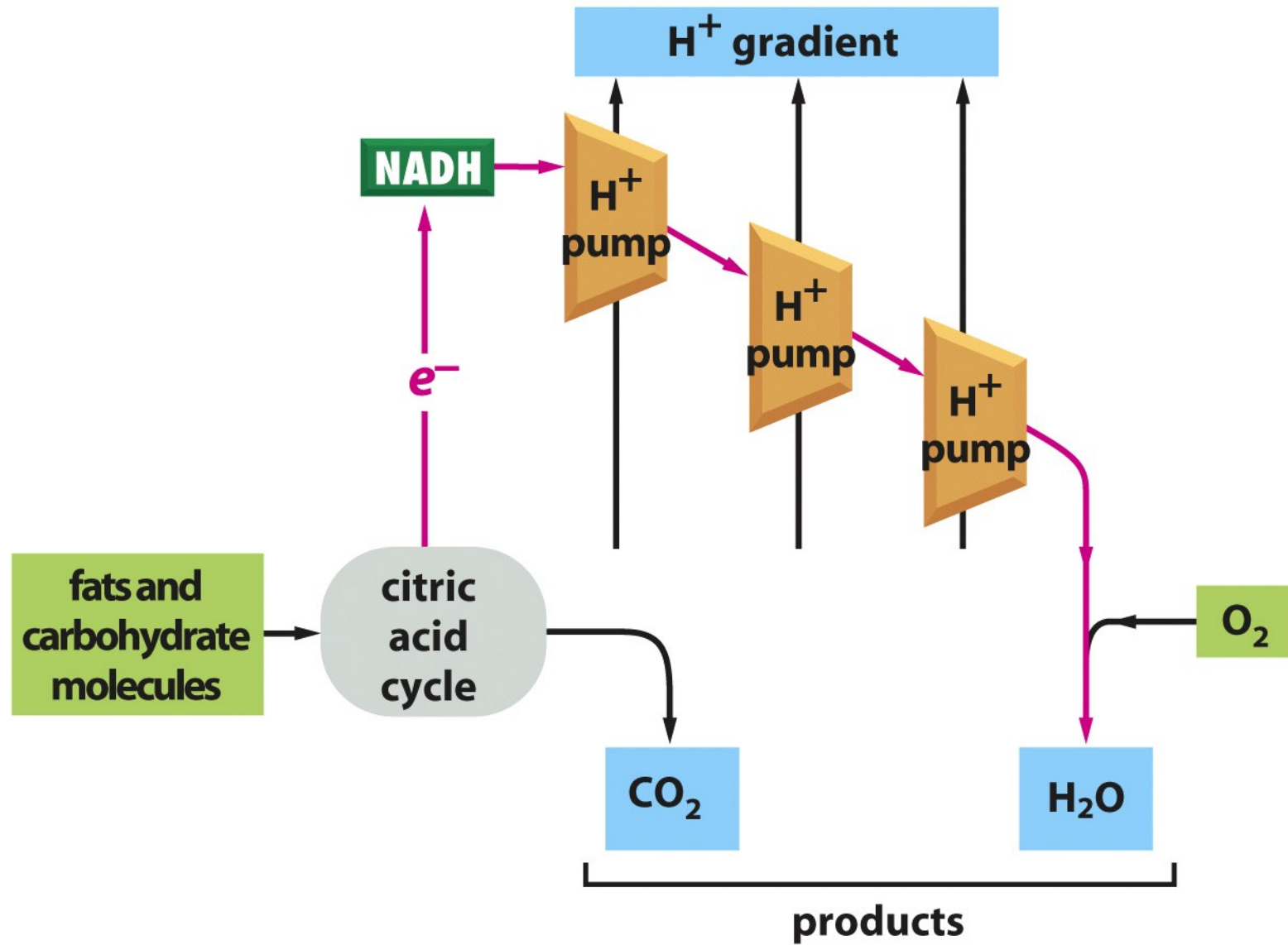


Figure 14-3a *Molecular Biology of the Cell* (© Garland Science 2008)

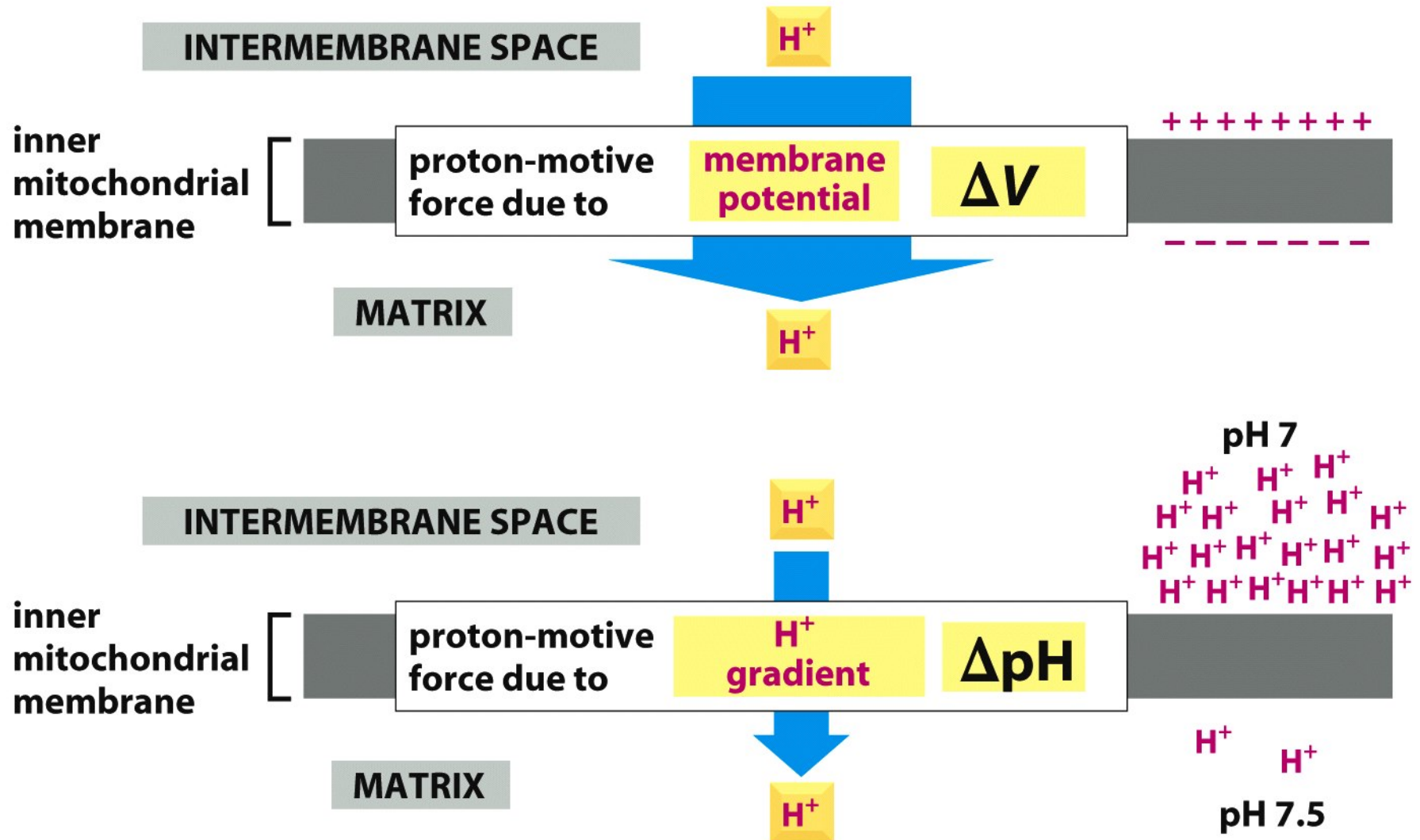


Figure 14-13 *Molecular Biology of the Cell* (© Garland Science 2008)

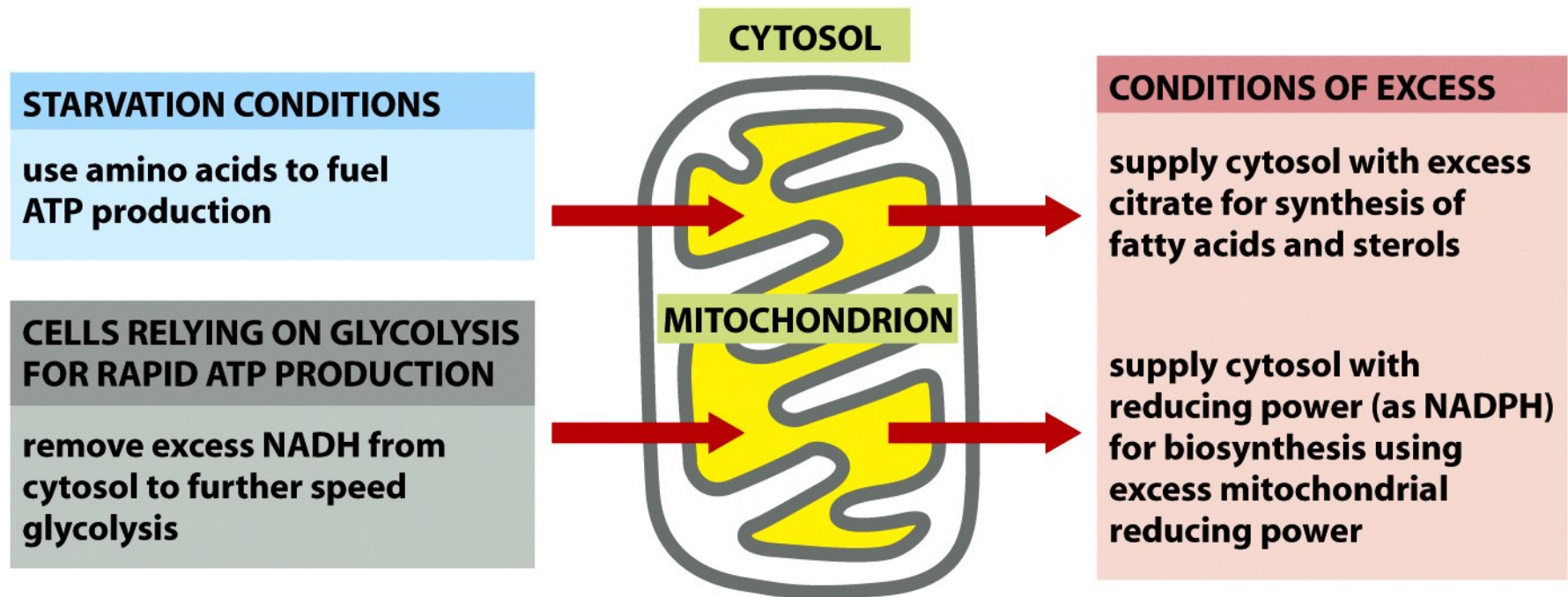
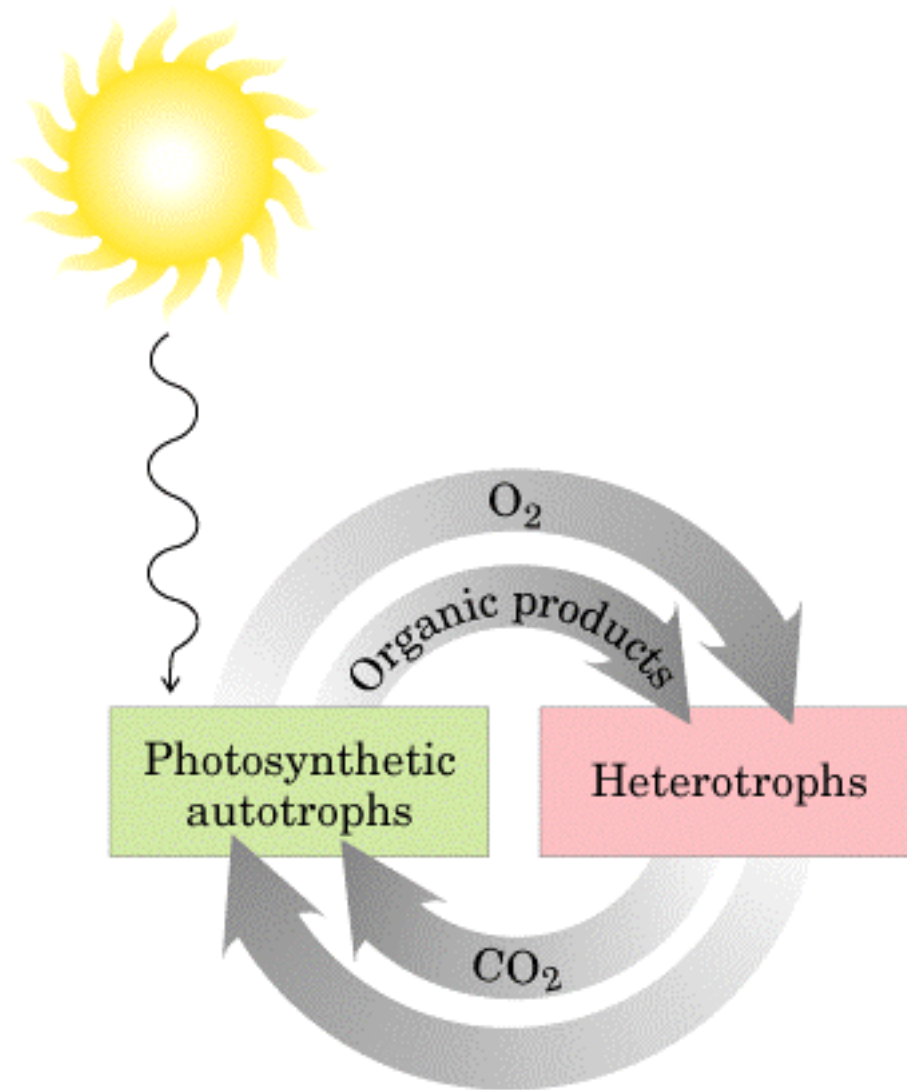


Figure 14-32 *Molecular Biology of the Cell* (© Garland Science 2008)



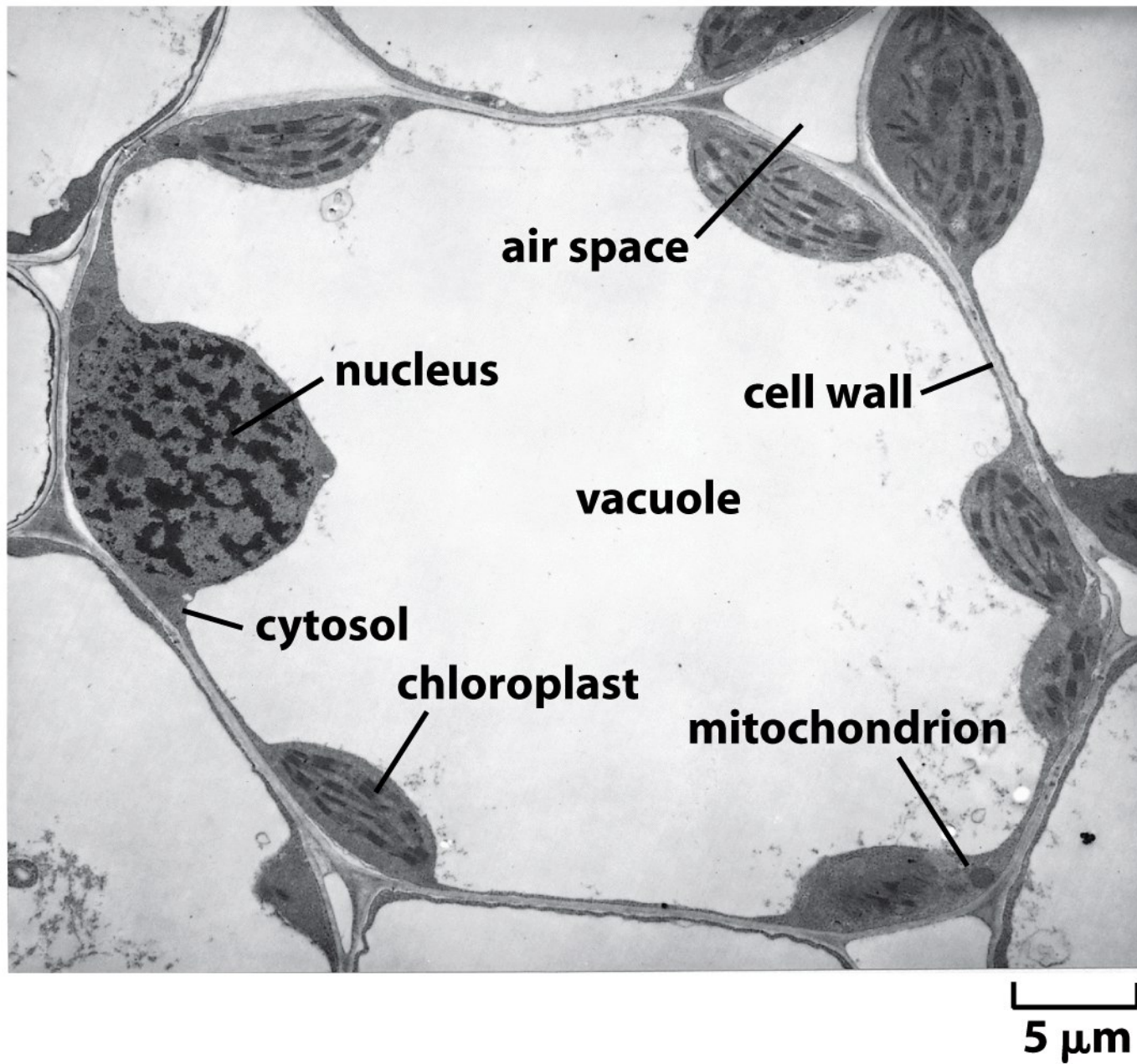


Figure 14-35a *Molecular Biology of the Cell* (© Garland Science 2008)

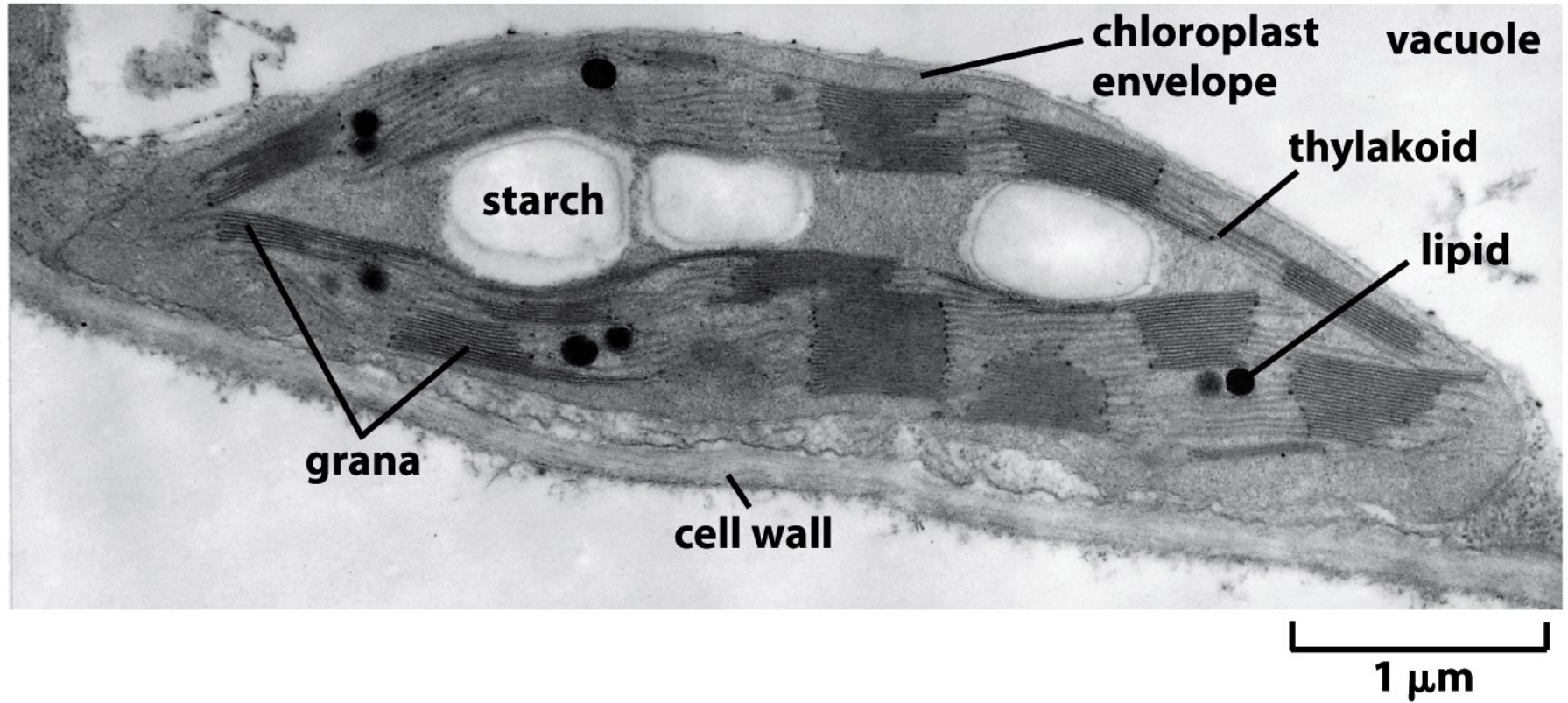


Figure 14-35b *Molecular Biology of the Cell* (© Garland Science 2008)

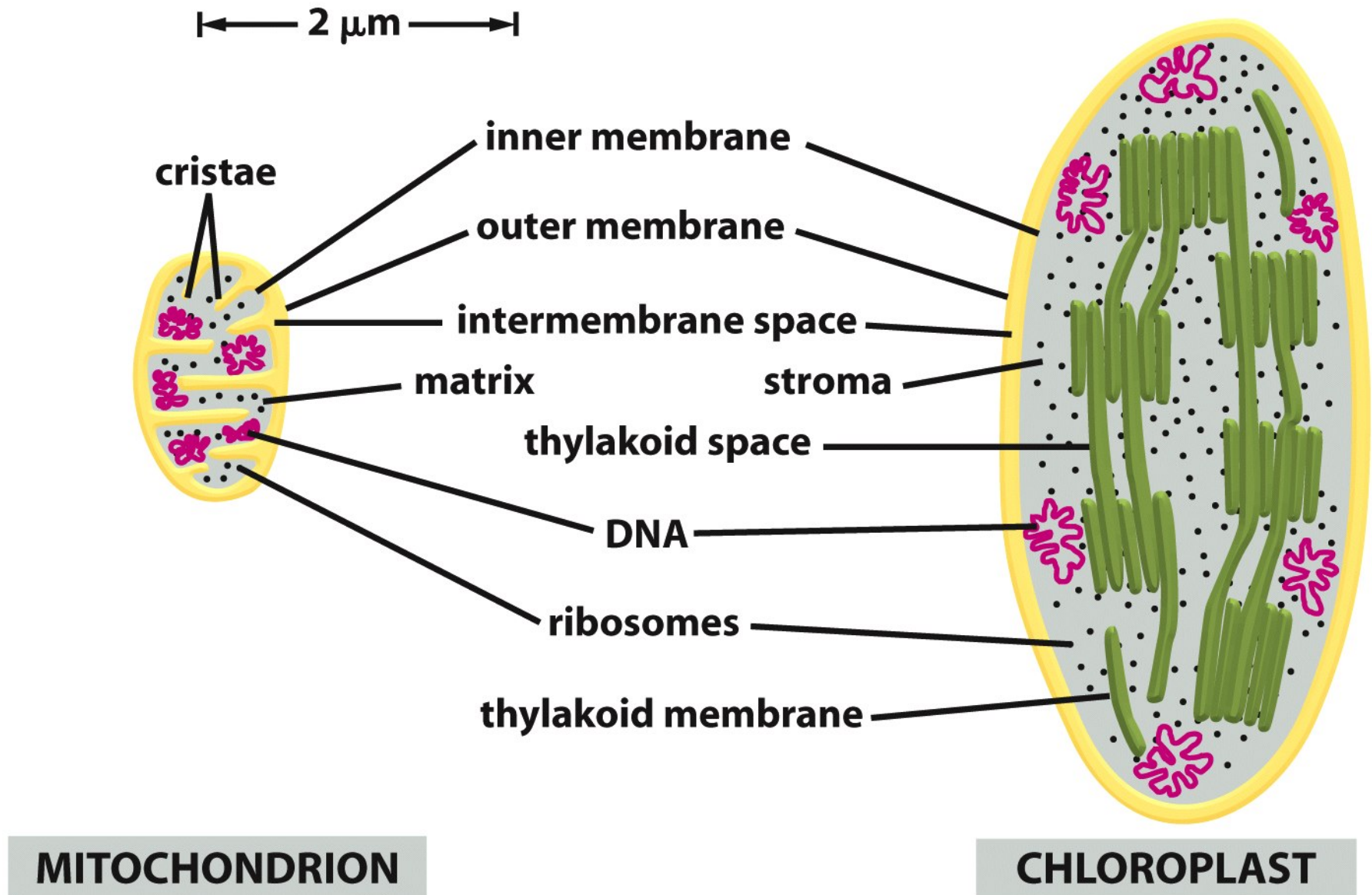


Figure 14-37 *Molecular Biology of the Cell* (© Garland Science 2008)

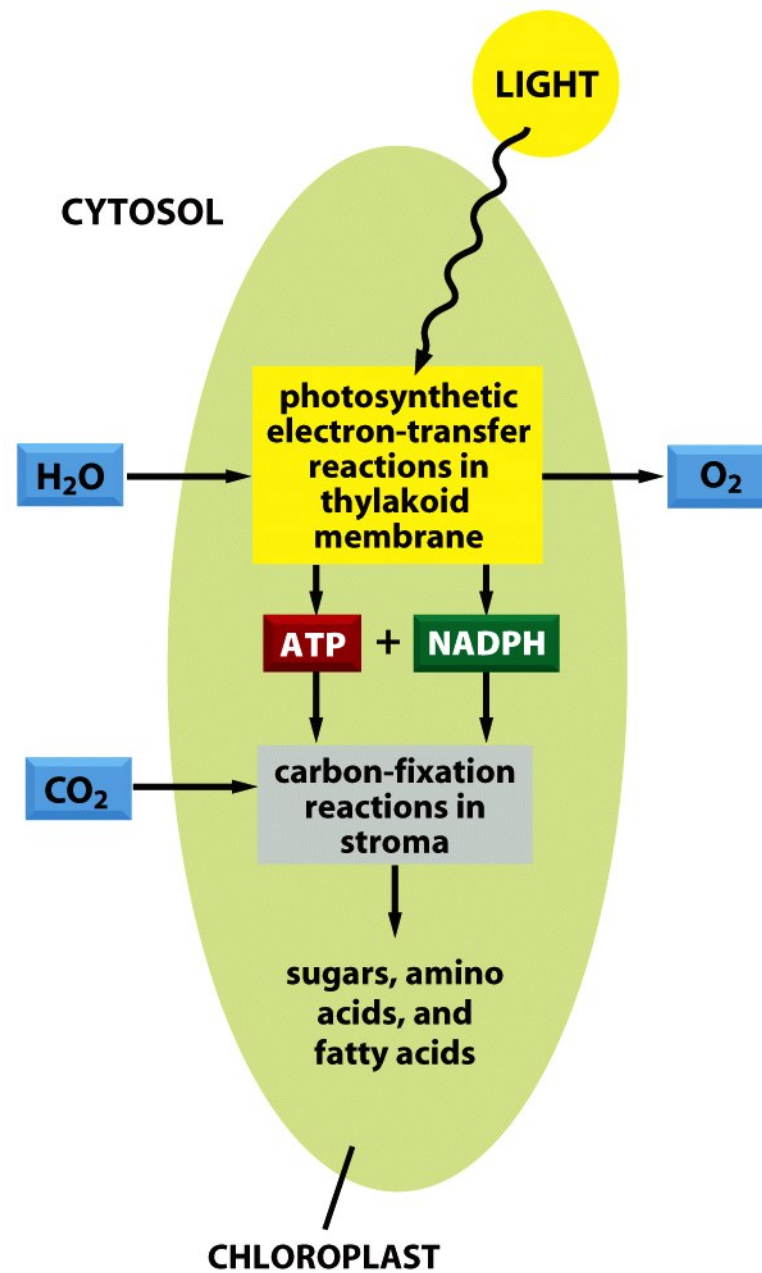


Figure 14-38 *Molecular Biology of the Cell* (© Garland Science 2008)

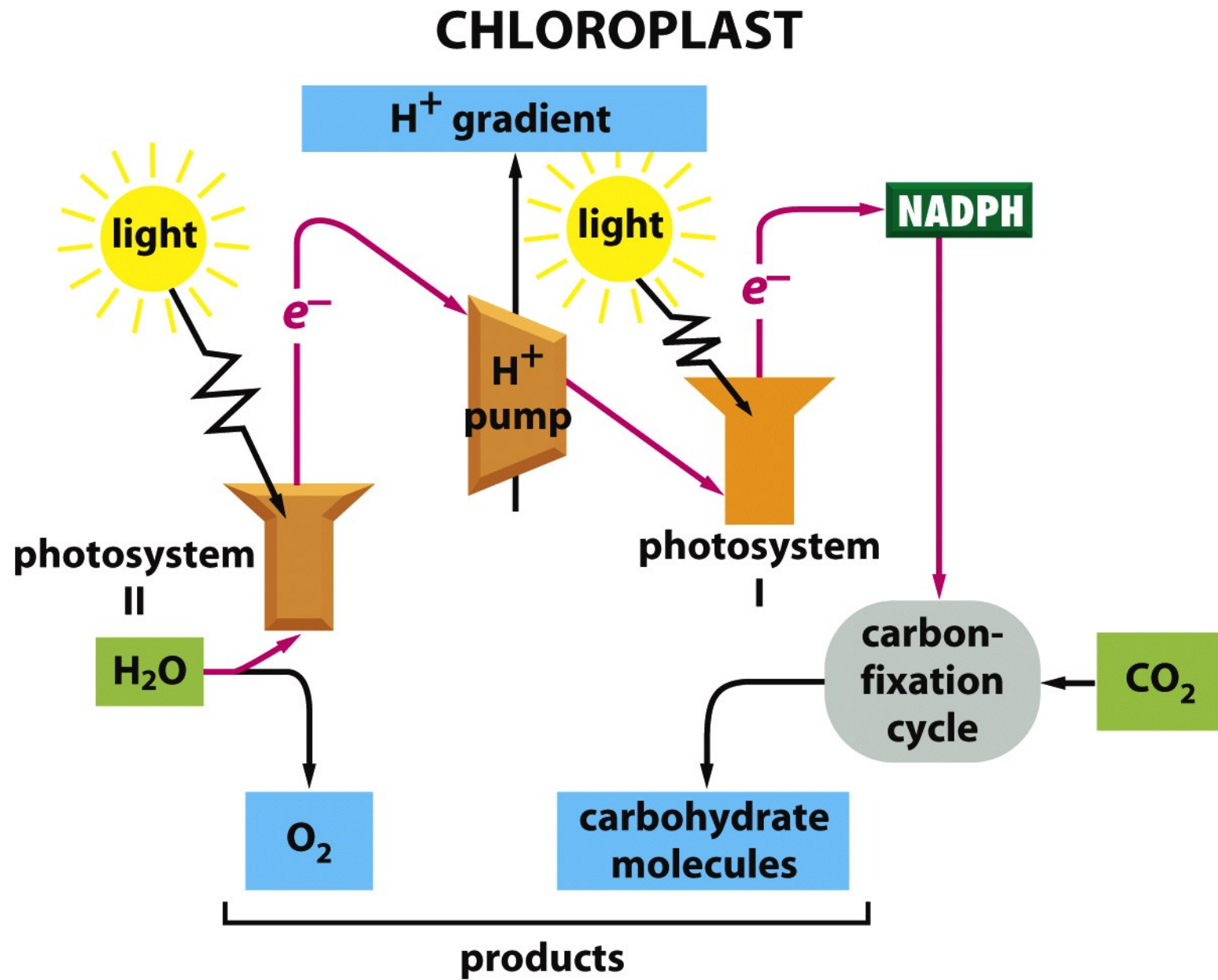


Figure 14-3b *Molecular Biology of the Cell* (© Garland Science 2008)

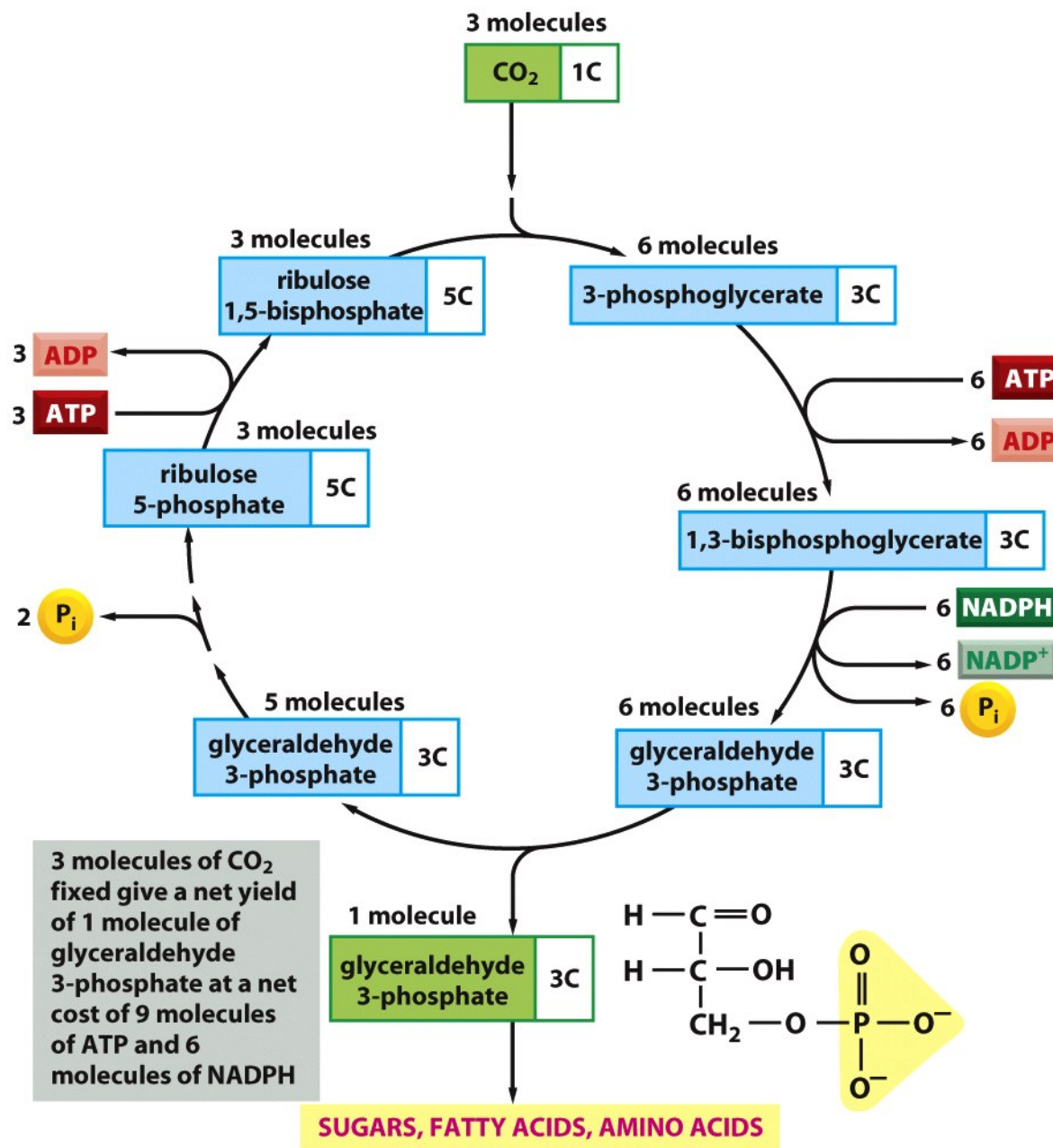


Figure 14-40 *Molecular Biology of the Cell* (© Garland Science 2008)

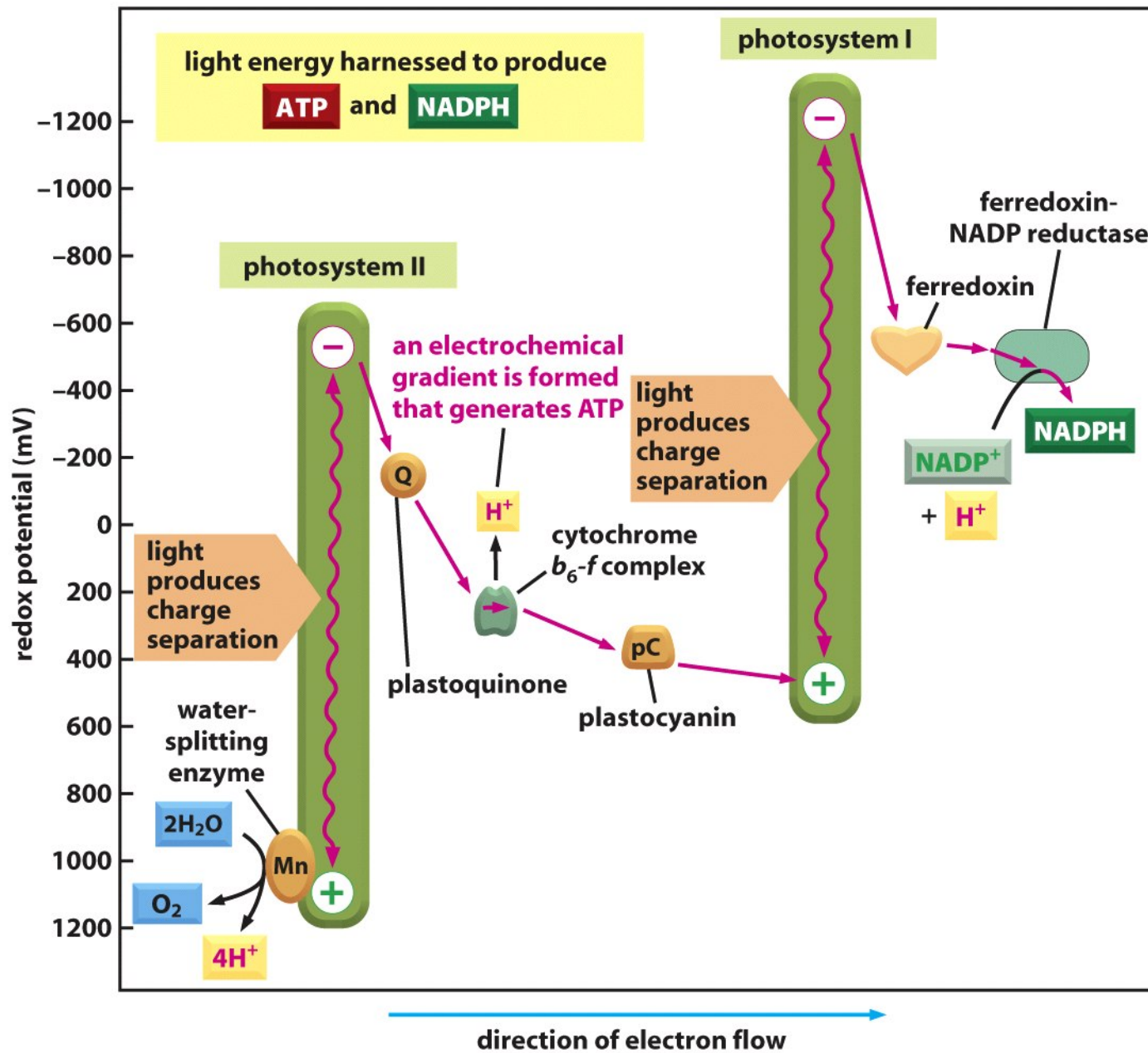


Figure 14-49 *Molecular Biology of the Cell* (© Garland Science 2008)

INSTRUCCIONES:

-Repase la clase

-Lea los siguientes artículos de wikipedia en inglés:

Lynn Margulis; Krebs cycle; Peter D. Mitchell; Calvin cycle; RuBisCO; Linus Pauling.

-Defina y explique:

Temperatura absoluta; energía libre de Gibbs (potencial químico); metabolismo; enlace covalente; catálisis (estado de transición; energía de activación); glicólisis; beta-oxidación de ácidos grasos; respiración celular; hipótesis quimiosmótica; fosforilación oxidativa; hipótesis endosimbionte; mitocondria; cloroplasto; fotosíntesis.

-Compare mediante una tabla, señalando diferencias y similitudes:

Respiración celular v/s Fotosíntesis; beta-oxidación mitocondrial v/s beta-oxidación peroxisomal; fase clara y fase oscura de la fotosíntesis.

-Dibuje, señalando partes y procesos: una molécula de ATP; una molécula de NADH; un diagrama de la respiración celular; un diagrama de la fotosíntesis.

-Persista y organícese, la prueba se acerca a pasos agigantados.