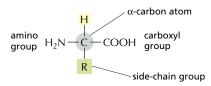
THE AMINO ACID

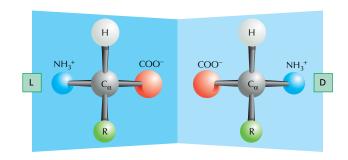
The general formula of an amino acid is



R is commonly one of 20 different side chains. At pH 7 both the amino and carboxyl groups are ionized.

OPTICAL ISOMERS

The α -carbon atom is asymmetric, which allows for two mirror image (or stereo-) isomers, L and D.



Proteins consist exclusively of L-amino acids.

FAMILIES OF AMINO ACIDS

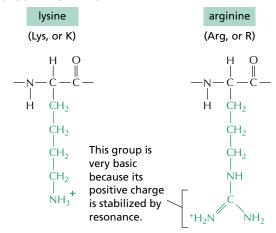
The common amino acids are grouped according to whether their side chains are

acidic basic uncharged polar nonpolar

These 20 amino acids are given both three-letter and one-letter abbreviations.

Thus: alanine = Ala = A

BASIC SIDE CHAINS



histidine (His, or H)

H O | || N-C-C-| | H CH₂

H CH₂
C
HN CH
HC=NH⁺

These nitrogens have a relatively weak affinity for an H⁺ and are only partly positive at neutral pH.

PEPTIDE BONDS

Amino acids are commonly joined together by an amide linkage, called a peptide bond.

Peptide bond: The four atoms in each *gray box* form a rigid planar unit. There is no rotation around the C–N bond.



Proteins are long polymers of amino acids linked by peptide bonds, and they are always written with the N-terminus toward the left. The sequence of this tripeptide is histidine-cysteine-valine.

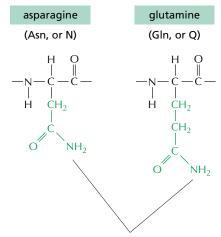
HC=

SH | CH2 | H | H | Carboxyl- or C-terminus | C | C | C | C | COO- | CH3 | CH3

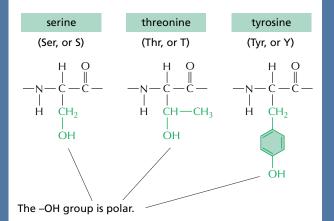
These two single bonds allow rotation, so that long chains of amino acids are very flexible.

ACIDIC SIDE CHAINS

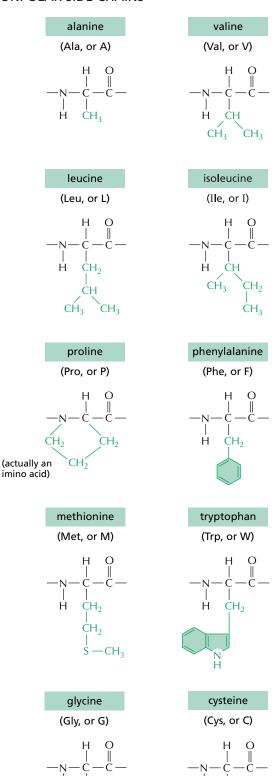
UNCHARGED POLAR SIDE CHAINS



Although the amide N is not charged at neutral pH, it is polar.



NONPOLAR SIDE CHAINS



Disulfide bonds can form between two cysteine side chains in proteins.

$$--CH_2-S-S-CH_2--$$